

## 166. The Configuration of Naturally Occurring Iridoid Glucosides at C(6) and C(8): A Complementary Assignment Aid by $^{13}\text{C}$ -NMR. Spectroscopy

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### Summary

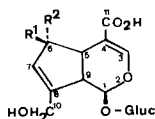
Evidence is presented which demonstrates that  $^{13}\text{C}$ -NMR. spectroscopy can be used with confidence in evaluating the configuration of  $\text{R}^1\text{R}^2\text{CHOH}$  centers at C(6) and C(8) of iridoid glucosides.

The configurational analysis with the help of NMR. carried out so far in this series used the proton [1]. This nucleus, however, suffers from two deficiencies: (i) proton spectra are complicated by spin-spin coupling which produces frequent overlap and renders spectrum interpretation difficult; (ii) the chemical shift differences of a proton in only slightly different environments are small and in the same order of magnitude as the long-range shielding effects. Thanks to the sensitivity of the  $^{13}\text{C}$ -chemical shifts to configurational changes, the method is ideally suited for distinguishing stereoisomers.

Several reports (*cf. e.g.* [2-4]) have appeared on the  $^{13}\text{C}$ -NMR. spectra of iridoid glucosides. However, no attempt has yet been made in this series to utilize the striking sensitivity of the  $^{13}\text{C}$ -chemical shifts to steric effects. We have recently reported the structure of some new ester iridoids by the aid of  $^{13}\text{C}$ -NMR. spectra [5] [6]. In order to establish a data basis for future structural studies and for throwing some light on the stereochemical dependence of  $^{13}\text{C}$ -chemical shifts, the  $^{13}\text{C}$ -spectra of three stereoisomeric pairs of iridoids are examined.

The  $^{13}\text{C}$ -NMR. spectral analysis of iridoid glucosides **1**, **2**, **3**, **4**, **5** and **6** gave the needed stereochemical information. The spectra studied here are presented in the *Table*. The assignments [7] of the  $^{13}\text{C}$ -NMR. signals were based on single frequency off-resonance decoupling technique, application of known chemical shift rules for hydroxy substitution and comparison among related compounds [8].

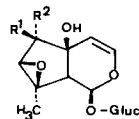
Inspection of the  $^{13}\text{C}$ -NMR. spectra of C(6)-epimers shows that on going from equatorial to axial substitution, *e.g.*, **1** to **2** or **3** to **4** a shielding ( $\gamma$ -effect) ( $\sim 3$  ppm) is observed for the C(4) whereas the C(8) has experienced an approximately equivalent deshielding. A  $\delta$ -effect of the substituent at C(6) exerted on C(3) in the



**1** Scandoside:

R<sup>1</sup> = OH; R<sup>2</sup> = H

**2** Desacetyl asperulosidic acid: R<sup>1</sup> = H; R<sup>2</sup> = OH

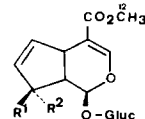


**3** Antirrinoside:

R<sup>1</sup> = OH; R<sup>2</sup> = H

**4** Procumbide:

R<sup>1</sup> = H; R<sup>2</sup> = OH



**5** Gardenoside:

R<sup>1</sup> = OH; R<sup>2</sup> = CH<sub>2</sub>OH

**6** Monotropein methylester:

R<sup>1</sup> = CH<sub>2</sub>OH; R<sup>2</sup> = OH

Table.  $^{13}\text{C-NMR}$ . Data of Iridoids<sup>a)</sup>

| C-Atom | 1      | 2      | 3      | 4      | 5                    | 6      |
|--------|--------|--------|--------|--------|----------------------|--------|
| 1      | 98.80  | 101.53 | 94.56  | 95.56  | 94.41                | 95.15  |
| 3      | 153.98 | 155.62 | 142.90 | 144.12 | 152.02               | 152.48 |
| 4      | 111.02 | 108.27 | 107.45 | 104.45 | 111.60               | 110.53 |
| 5      | 46.97  | 42.57  | 74.52  | 80.01  | 38.92                | 38.80  |
| 6      | 82.38  | 75.33  | 77.29  | 77.69  | 135.80 <sup>b)</sup> | 137.54 |
| 7      | 129.92 | 129.89 | 66.01  | 65.75  | 135.90 <sup>b)</sup> | 133.65 |
| 8      | 147.31 | 151.32 | 64.18  | 66.83  | 86.34                | 85.89  |
| 9      | 45.94  | 45.73  | 52.72  | 52.88  | 52.40                | 45.42  |
| 10     | 61.06  | 61.61  | 17.53  | 17.59  | 67.17                | 68.26  |
| 11     | 172.07 | 170.92 | -      | -      | 169.69               | 168.97 |
| 12     | -      | -      | -      | -      | 51.62                | 51.58  |

<sup>a)</sup> The spectra were recorded in  $\text{CD}_3\text{OD}$ . Chemical shifts in ppm relative to internal  $(\text{CH}_3)_4\text{Si}$ . Additional signals arising from glucose. <sup>b)</sup> These values could be interchanged.

C(6)-epimers, although of lower extent ( $\sim 1.5$  ppm), is observed. The shielding effects can be rationalized on the basis of electric field effects [9] and find precedence in other systems [10]. It is of interest to point out the different chemical shift values for C(6) associated with the equatorial (1) and axial (2) hydroxyl function at this carbon atom.

Comparison of the C(1)-, C(6)-, C(7)- and C(9)-chemical shifts of one of the C(8)-epimers with those of the other distinguish the two compounds easily and reveal their C(8)-configurations as depicted in stereostructures 5 and 6. Since the cyclopentane ring is frequently substituted, shifts caused to C(7) and C(6) with a change in configuration at C(8), may not be useful. It has recently been shown [4] [11] that changes in substituent or configuration at C(7) have little influence on the chemical shift value of C(9). Thus, the C(9)-chemical shift of C(8)-epimers is a diagnostic constant for revealing the C(8)-configuration of such natural products [11]. The shift induced on the  $\beta$ -carbon atom C(9) with the change of the configuration at C(8) can be rationalized on the basis of a  $\text{C}_\beta$ -electron-cloud distortion by the fluctuating  $\text{C}_\alpha$ -OH dipole [10].

These findings show clearly that  $^{13}\text{C-NMR}$ . spectroscopy provides a useful sensitive method for determining the configuration of  $\text{R}^1\text{R}^2\text{CHOH}$  centers at C(6) and C(8) in iridoids.

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